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1. Match the equation type on the left to its representation on the right. synthesis(a) AX BY AY BX decomposition(b) A BX AX B single-displacement(c) A B AX double-displacement(d) AX A X. 2. In the reaction described by the equation 2Al(s) + 3Fe(NO₃)₃(aq) → 2Al(NO₃)₃(aq) + 3Fe(s), the aluminum is oxidized and the iron is reduced.

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The formation of water according to the equation 2H₂(g) + O₂(g) → 2H₂O(l) shows that 2 molecules (made of 4 atoms) of hydrogen and 1 molecule (made of 2 atoms) of oxygen produce 2 molecules of water. The total mass of the product, water, is equal to the sum of the masses of each of the reactants, hydrogen and oxygen.

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